

CHEMISTRY

Examination Board

OCR (A)

Year 12 Course

Content Overview

- Module 1 Development of practical skills in chemistry
- Module 2 Foundations in chemistry
- Module 3 Periodic table and energy
- Module 4 Core organic chemistry
- Two mock exams at the end of Year 12, both covering content from all 4 modules.

Year 13 Course

Content Overview

- Module 5 Physical chemistry and transition elements
- Module 6 Organic chemistry and analysis

Assessment Overview

- Three official exams at the end of Year 13:
- Periodic table, elements and physical chemistry (Paper 1) assesses content from modules 1, 2, 3 and 5.
 - 100 marks 2 hours and 15 minutes written paper 37% of total A level
- Synthesis and analytical techniques (Paper 2) assesses content from modules 1, 2, 4 and 6.
 - 100 marks 2 hours and 15 minutes written paper 37% of total A level
- Unified chemistry (Paper 3) assesses content from all modules (1 6).
 70 marks 1 hour and 30 minutes written paper
 26% of total A level

Note that the Year 12 exams do not contribute to the final A level mark and therefore 2 years of content is assessed in the Year 13 exams.

Coursework

The A Level Sciences no longer have a traditional coursework component. You will have to complete 12 practical techniques and you will be assessed on these in class and via the written exams. Students will receive a 'pass' in the Chemistry Practical Endorsement if they can demonstrate that they are consistently competent in all 12 practical techniques. The Practical Endorsement is usually an entry requirement for science degrees.

Useful websites / reading materials

Students are given checklists of the topic content at the end of each unit and of the whole course content towards the end of each year.



During lessons students will have access to the course textbook, **shown in bold type** below.

It is strongly recommended that students purchase this textbook for their own use at home (and in lessons if they prefer). They can also personalise their learning by annotating the text using highlighter pens, spider diagrams, cross referencing to their own written notes.

Class textbooks:

A level Chemistry A for OCR Year 1 and AS, OUP Oxford, ISBN 978-0-19-835196-2 A level Chemistry A for OCR Year 2, OUP Oxford, ISBN 978-0-19-835765-0

(Also available A level Chemistry for OCR, OUP Oxford, ISBN 978-0-19-835197-9 which is both of the above in the same book)

There are other resources available for students to help them with their study of chemistry:

- The Resources Centre has a selection of A Level Chemistry textbooks for students to consult, either to assist their understanding, or to broaden their knowledge. Students may also find books of interest at their local library.
- Periodicals such as New Scientist or Scientific American
- The school receives a package of journals aimed at both staff and students from the Royal Society of Chemistry. Students will probably not wish to read these from cover to cover but they are a valuable reference source and some of the articles are fascinating! They can use ideas that they have read about to include in the UCAS personal statement.
- The Internet has huge amounts of chemical information; we can help with suggestions of suitable websites but there are many more waiting to be discovered!
 The most important website is the OCR course website. They can download the specification for the course; past papers and mark schemes.

http://www.ocr.org.uk/qualifications/type/gce/science/chemistry_a/

Other websites include:

www.chemguide.co.uk www.physicsandmathstutor.com http://www.rsc.org/Chemsoc/

Recommended study

• Approximate 5 hours of Chemistry per week, increasing when there are examinations to prepare for.

Independent Study:

- Reading over and writing up notes in neat
- Completing set homework tasks
- Answering and self-assessing past paper questions
- Preparing for coursework (practice graphs, tables, data analysis etc)
- Watching science programmes on TV / online
- Reading ahead for the next lesson
- Small group revision and study sessions